S		UNIVE	RSITY OF EAST								
		Study programm	ne: Chemical Engl	nology							
		Cycle I	Cycle I								
Course title Instrum		mental Methods									
Department		Depa	rtment for Chemis	tment for Chemistry– Faculty of Technology Zvornik							
Course code			Coι	Course status		ster	ECTS				
04	-1-021	-4	Co	Compulsory			5				
Teacher Zoran Obre		enović, PhD, Asso	ociate Professor								
Teaching assistant		Milomirka (Obrenović, MSc,	prenović, MSc, Senior Teaching Assitant							
Number of classes/ teachin week)		teaching	workload (per	orkload (per Individual stu		in hours per	Student workload coefficient S₀				
Lectures	Lectures Au		Laboratory exercises	Lectures	Auditory exercises	Laboratory exercises	S₀				
2	0*45	0	2	45	45	0	1.50				
	Z^15	+0*15+2*15	Total course w	orklaad 60 ± 00 -	(2°15°1.50 +)	<u>0″15°1.50 + 2*</u> mester	15"1.50)=90 hours				
		After finish	ing the course, st	udents will be able	e to:	11103101					
Learning outcomes		 demonstrate and utilize the knowledge of the principles and procedures of instrumental analysis apply the appropriate instrumental analytical procedure and apparatus for the necessary analytical testing conduct correct sampling, prepare the sample, and perform the instrumental analytical procedure correctly display the results of the analysis in the prescribed units. 									
Prerequisites N		None.	None.								
leaching meth	ods	Lectures, experimental exercises, calculations, mid-term tests (colloquia), consultations.									
Syllabus out per week	lline	 Instrumental methods in modern analytical chemistry, Preparation of representative samples Measurement, accuracy, precision, types and causes of errors, basic statistical processing of results Ion exchange resins, ion exchange balance, application of ion exchangers in analytics; Chromatography, theoretical basis, division of chromatographic methods Application of chromatography in analysis, gas, liquid, and paper chromatography Colorimetry and photometry, theoretical basis, typical colorimeters and photometers and application in determinations Atomic absorption spectroscopy (AAS), theoretical basis and Mid-term test 1 Apparatus for AAS, application in modern analytics, advantages and limitations Polarimetry, polarized light, properties and regularities of the phenomenon of rotation of the plane of polarized light; polarimeters, construction and analytical application Polarimetry, polarized light, properties and laws of occurrence of plane rotation of polarized light; polarimeters, construction and analytical application Thermometry, differential thermal analysis, thermogravimetric analysis, thermometric titration Electroanalytical methods, classification and application in determinations, Principles of conductometry Direct conductometry, conductometric titration Potentiometry, theoretical basis, indicator and reference electrodes Measurement of rN-value, ion-selective electrode and rX-metry Mid-term test 2 									
Δι	thor			Title public	ading	Year	Pages				
Станојевић. Д.		Д.	Аналитичка х	Аналитичка хемија, "Српска књига". Рума-Е		рад 2004	173-261				
Мишовић, Ј., Аст, Т.		Инструмента	Инструменталне методе хемијске анализе, ТМФ, Београд		IΦ, 2000	9-205					

Additional reading											
Author		Title, publisher	Year	•	Pages						
Фотић, Љ).	Инструменталне методе хемијске анализе, Практикум за вежбе, ТМФ, Београд			-						
Gunzler, H., Williams, A.		Handbook of analytical technics, Wiley VCH 200			173-323						
Harvey, D.		Modern Analytical chemistry, The McGraw Hill Co, New York	2000)	461-615						
		Type of student evaluation	Grade points	Percentage							
Obligations	Pre-exam obligations										
Obligations,		Atter	6	6 %							
methode and		Laboratory exe	36	36 %							
aradina system		Two mid-term tests -	14+14	28 %							
grading system	Final examination										
		Final examination (oral/v	30	30 %							
	Total			100	100 %						
Web page	www.tfzv.ues.rs.ba										
Date	2023										