SUCTORY			UNIVE	ALC VICTORIO						
			Study	ra nrogramn						
San Charles			Sludy							
Course title		Bioch	omist		3 450 10 Cart					
Department	Dioci	Department for Food technology – Faculty of Technology Zvornik								
		1 - 0 - 0								
Course code		de	C		urse status	Seme	ester	ECTS		
04-1-100-		5			Compulsory			7		
Teacher N		Milenko Sr	enko Smiljanić, PhD, a		ociate professor					
assistant		Milenko Sr	nko Smiljanić, PhD, associate professor							
Number of hours/ tead week)		eaching	hing workload (per		Individual s	udent workload semester)	(in hours per	Student workload coefficient S₀		
Lectures	Lectures Aud		ry Laboratory		Lectures	Auditory	Laborator	y S₀		
3	exe	1	s exercises		60	20	40	1.33		
Ŭ	3*15+1*15		=90 h	ours		(3*15*1.33+	-1*15*1.33+2*	15*1.33)=120 hours		
			Tota	al course wo	orkload 90 + 120	=210 hours per s	emester			
Learning outcomes		After finishing the course, students will be able to: 1. demonstrate and utilize the necessary level of knowledge in biochemistry required for the study of food science and food technologies 2. demonstrate and utilize basic knowledge about biochemical compounds (ability to understand the structure and biological function of proteins, enzymes and coenzymes, nucleic acids, carbohydrates and lipids) and reactions (biosynthesis and breakdown of biomolecules) that form the basis of the main life processes. 3. demonstrate and utilize the knowledge of the main pathways and energy balances of metabolic transformations in biomolecules 4. master the basics of the regulation of the appropriate metabolic pathways 5. understand the integration of biochemical transformations of natural compounds with energy transformations in living organisms 6. demonstrate and utilize the basic knowledge needed to understand experimental methods based on the biological activity of molecules 7. demonstrate and utilize the skills of performing simple biochemical experiments, presenting literature data and experimental results, effective learning, critical thinking and evaluation of teaching and learning outcomes.								
Teaching metr	_ectures, auditory and laboratory exercises, mid-term tests (colloquia).									
Syllabus ou per week	1. 5 2. 6 3. 6 4. 6 5. 6 6. 6 7. 6 8. 6 9. 7 10. 5 11. 6 12. 1 13. 1 main 14	<ol> <li>Basics of biocatalysis and biochemistry of functional compounds (catalysis in biological systems, structure, classification, mechanism and specificity of enzyme action, enzyme inhibition, cofactors).</li> <li>Influence of individual factors on enzyme activity (environmental reaction, temperature, enzyme concentration, substrate concentration), regulation of enzymatic reactions.</li> <li>Biochemical properties of carbohydrates and important ways of breaking down carbohydrates (anaerobic and aerobic decomposition).</li> <li>Oxidative decarboxylation of pyruvic acids, Krebs cycle.</li> <li>Carbon dioxide assimilation (photosynthesis, heterosynthesis, chemosynthesis).</li> <li>Biochemical properties of lipids and their catabolism. Knowledge test.</li> <li>Biosynthesis of fatty acids, triglycerides and phospholipids.</li> <li>Amino acids, properties, general reactions of amino acid metabolism.</li> <li>Structure, properties and functions of nucleic acids.</li> <li>Protein biosynthesis.</li> <li>Vitamins and minerals.</li> <li>Metabolism as a unique system. Connection, control and regulation of the metabolism of the main groups of natural compounds.</li> </ol>								

membranes, transport of ions in the cell. 15. Methods of analytical biochemistry. Knowledge test. Final test. Mid-term tests are taken after the 7th week and the 15th week. Semester verification is required after the 15th week.												
Obligatory reading												
Author		Title, publisher	Year	r Pages								
Величковић	рД.	Основи биохемије за студенте биотехничких наука, Универзитет у Београду, Београд.	2000	0								
Karlson P		Biokemija. Školska knjiga, Zagreb.	1993									
Бараћ М., С. Стано Пешић, Д. Зс	ојевић, М. орић	Практикум из биохемије, Универзитет у Београду, Београд.	2010									
Џамић М		Практикум из биохемије, Грађевинска књига, Београд.	1986									
		Additional reading										
Author		Title, publisher	Year	Pages								
Jeremy M. Berg, Tymoczko, Luber	John L. t Stryer.	Biochemistry, 5th edition, W. H. Freeman and Company, New York.	2002									
Nelson,D., Co	х, М.	Lehninger Principles of Biochemistry, fourth edition, Freeman, W. H. & Company, New York.	2004									
		Type of student evaluation	Grade points	Percentage								
	Pre-exam obligations											
Obligations,		Atten	6	6 %								
assessment		Mid-tern	22	22 %								
methods and		Mid-term	22	22 %								
grading system	Einel aversit	Laboratory exe	20	20 %								
		Iduoii Einal oversingtion	30	30.0/								
	Total		100	100 %								
Web nage												
Data	2002											
Date	2025											